SYNTHESIS AND PHOTOPHYSICAL PROPERTIES OF PHOSPHINE ACETYLIDE AMIDINATE STABILIZED COPPER(I) AND GOLD(I) HETEROBIMETALLIC COMPLEXES

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Here, we present an unprecedented synthesis of phosphine acetylide amidinate stabilized copper(I) and gold(I) heterobimetallic complexes. The synthesis starts with the preparation of ligand $[{Ph_2PC=CC(NDipp)_2}Li(thf)_2]$ (1) by the reaction of Ph₂PC=CLi and Dipp carbodiimide. Subsequently, salt metathesis reactions with CuCl and Au(tht)Cl was carried out, leading to the formation of $[{Ph_2PC \equiv CC(NDipp)_2}_2Cu_2]$ (2) and $[{Ph_2PC \equiv CC(NDipp)_2}_2Au_2]$ (3), respectively [1]. These two complexes exhibit distinct molecular structures due to different coordination sites of ligand. Exploiting the efficiency of compound 2 as a metalloligand, we further synthesized Cu(I) and Au(I) containing heterobimetallic complexes by harvesting the coordination potential of the phosphine. This led to the formation of complexes of the type $[{(AuX)Ph_2PC = CC(NDipp)_2}_2Cu_2]$ (X = Cl (4), Br (5), I (6) and Mes (7)) by reacting 3 with 2 eq. of CuX (X = Cl, Br, I, and Mes). Interestingly, these complexes, 4-7, could also be obtained through a convenient one-pot reaction involving 1, Au(tht)Cl, and CuX in high yield as compared to above stepwise procedure. Compounds 2 and 4-7 are yellow coloured and they exhibit bright yellow coloured luminescence both in solid state and in solution under UV light. They show phosphorescence emission with a maximum between 500-620 nm in solid state and between 530-650 nm in DCM solution.



Figure 1: Synthesis schemes of Cu(I) and Au(I) heterobimetallic compounds

^[1] Feuerstein, T. J.; Seifert, T. P.; Jung, A. P.; Müller, R.; Lebedkin, S.; Kappes, M. M.; Roesky, P. W. Efficient Blue Phosphorescence in Gold(I)-Acetylide Functionalized Coinage Metal Bis(amidinate) Complexes *Chem. Eur. J.* 2020, 26, 16676–16682.