## ELUCIDATING THE REACTIVITY OF EARLY TRANSITION METAL COMPLEXES BEARING ELECTRON POOR Cp-LIGANDS

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Electron poor Cp-ligands are still a neglected ligand class. We have prepared imidazole-ylidene derived fulvalenes that can be considered  $\pi$ -isoelectronic to sesquifulvalene, and which are better described as imidazolium substituted cyclopentadienides due to their highly ylidic character [1].

Upon substitution with additional imidazolium moieties, we achieved several Cp-ligands whose electron donating character can be adjusted incrementally depending on the number of imidazolium moieties and the coplanarity of the Cp and the imidazolium moieties [2].

The versatility of the ligands lies in the straightforward formation of Cp complexes by mere addition to the metal center, avoiding any salt formation or redox reactions [3].

We have now investigated the formation of Group 3 [4] and Group 4 [5] metal complexes with these (Imi-Cp)-ligands and found the formation of interesting equilibria, which we elucidated by NMR techniques, and which depend strongly on the reaction conditions, e.g. solvent and counter ions.

+ 
$$[ScCl_3(thf)_3]$$
 +  $3 thf$  +  $3$ 

As the complexes contain electron poorer metal centers than regular Cp complexes, they should be interesting for Lewis-acid catalyzed reactions. Therefore, we investigated (Imi-Cp) titanium complexes in the intramolecular hydroamination of terminal alkenes, whose results will also be discussed in the presentation [5].

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<sup>[2]</sup> a) F. Mazzotta, G. Zitzer, B. Speiser, D. Kunz, Chem. Eur. J. 2020, 26, 16291–16305; b) F. Mazzotta, K. W. Törnroos, D. Kunz, Organometallics 2020, 39, 3590–3601.

<sup>[3]</sup> D. Schmid, A. Seyboldt, K. Eichele, D. Kunz, Dalton Trans. 2017, 46, 29–32.

<sup>[4]</sup> F. Mazzotta, D. Kunz, Organometallics 2021, 40, 3003-3011.

<sup>[5]</sup> unpublished results.