THEORETICAL STUDY OF THE SEMIHYDROGENATION OF ALKYNES CATALYZED BY IMIDAZOLYL AMINO MOLYBDENUM CLUSTER SULFIDES

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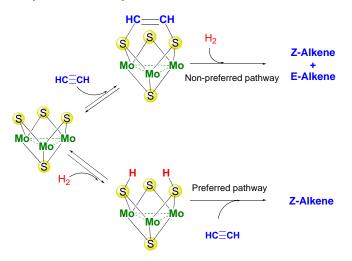
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We have recently reported a novel protocol for the semihydrogenation of alkynes using the cluster $[Mo_3S_4Cl_3(ImNH_2)_3]^+$ (1⁺) and selectively affording the (Z)- alkenes with excellent yields [1]. Mechanistic control experiments suggested a sulfur-based mechanism. While the reaction with diphenylacetylene (dpa) takes place, the presence of a dithiolene $Mo_3(\mu_3-S)(\mu-S)(\mu_3-SC(Ph)=C(Ph)S)$ (2⁺) adduct formed due to the bridging sulfur atoms of this cluster interacting with dpa was noticed. This adduct disappears with the time, and at the end of the reaction only 1⁺ remains.

Herein we report a DFT-based theoretical study on the semihydrogenation of dpa catalyzed by $\mathbf{1}^+$. Two pathways have been explored. One of them begins with the formation of $\mathbf{2}^+$, followed by its hydrogenation to render an unstable intermediate ($\mathbf{3}^+$) that can afford the Z-alkene. However, isomerization of $\mathbf{3}^+$ opens additional pathways leading either to the Z- as well as the E-alkene.

The other pathway begins with the hydrogenation of two bridging sulfurs of $\mathbf{1}^+$ obtaining a hydrogenated cluster $\mathbf{4}^+$, as already described by us [2]. Thereafter, $\mathbf{4}^+$ reacts with dpa in a very smooth way, including intersystem crossings with the triplet electronic state, to render solely the Z-alkene.

Although the formation of 2^+ is preferred over the formation of 4^+ , the large energy needed to follow up from 2^+ prevents this pathway to take place, and the system evolves via the second pathway to selectively render the Z-alkene.



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^[2] Guillamón, E., Oliva, M., Andrés, J., Pedrajas, E., Safont, V. S., Algarra, A. G., Basallote, M. G. ACS Catalysis 2021, 11, 608-614